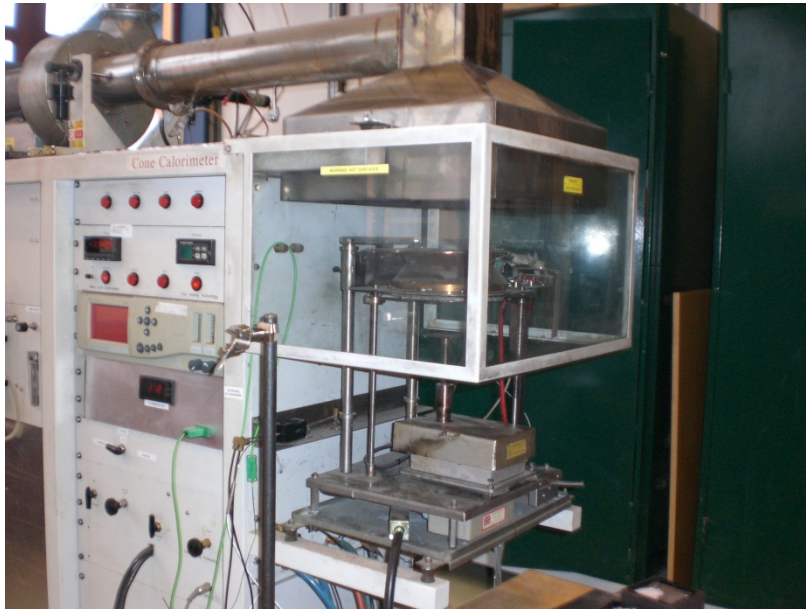
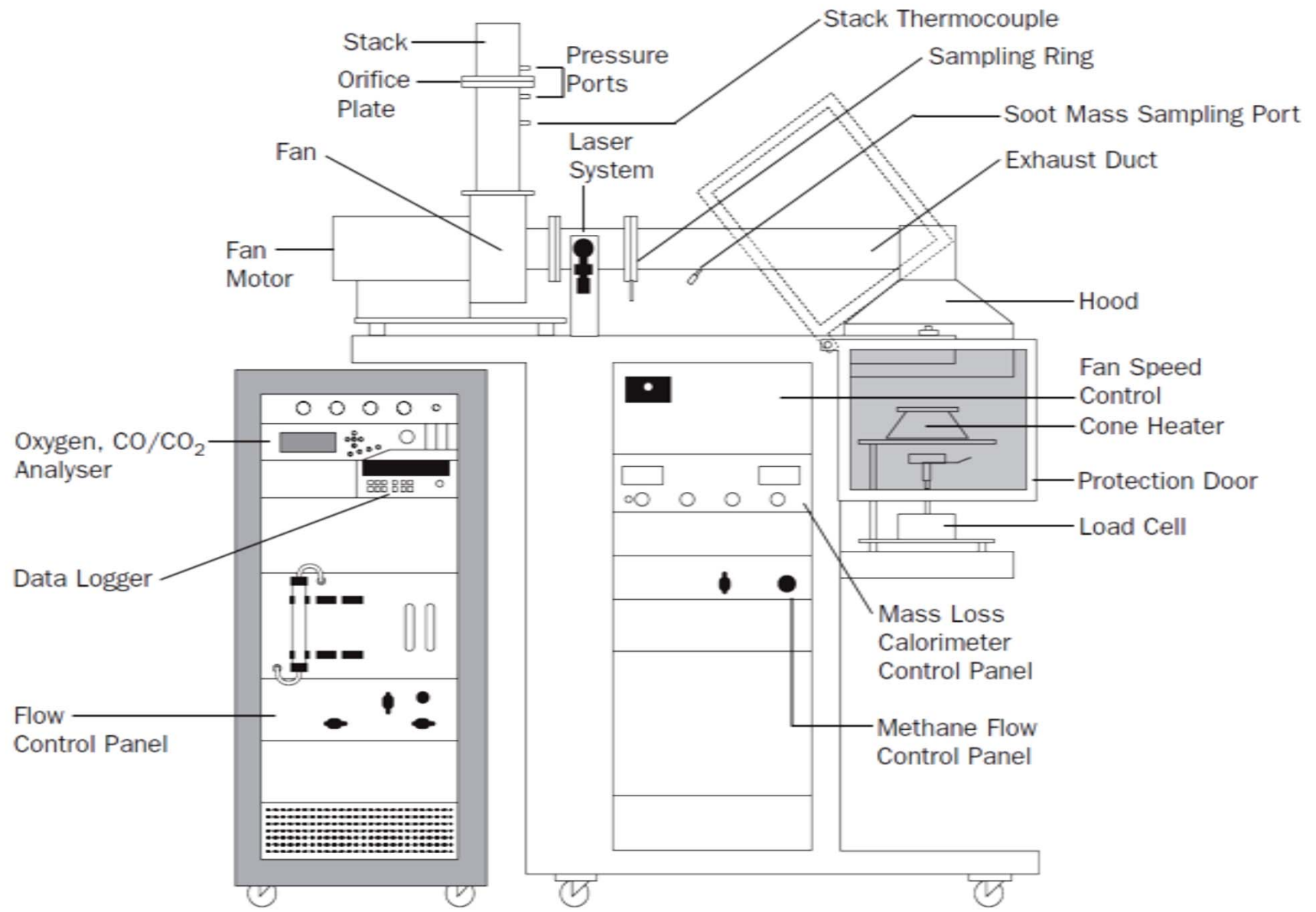
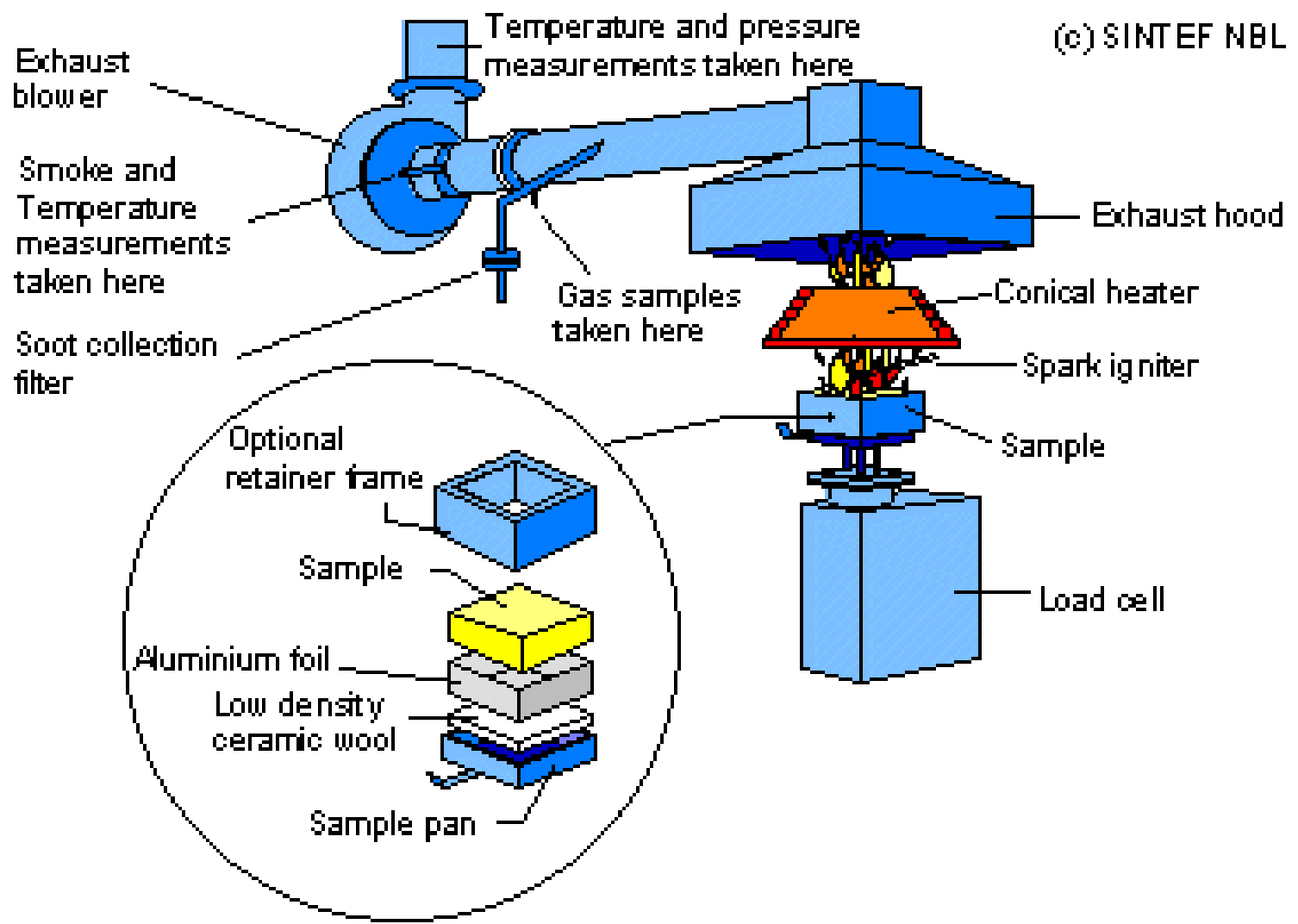


# Cone Calorimeter







# Fundamental Issues

- Heat Release Rate is obtained indirectly by measuring O<sub>2</sub> consumption
- Mass loss rate – real time by load cell
- Incident heat flux of 0 to 100 kW/m<sup>2</sup>

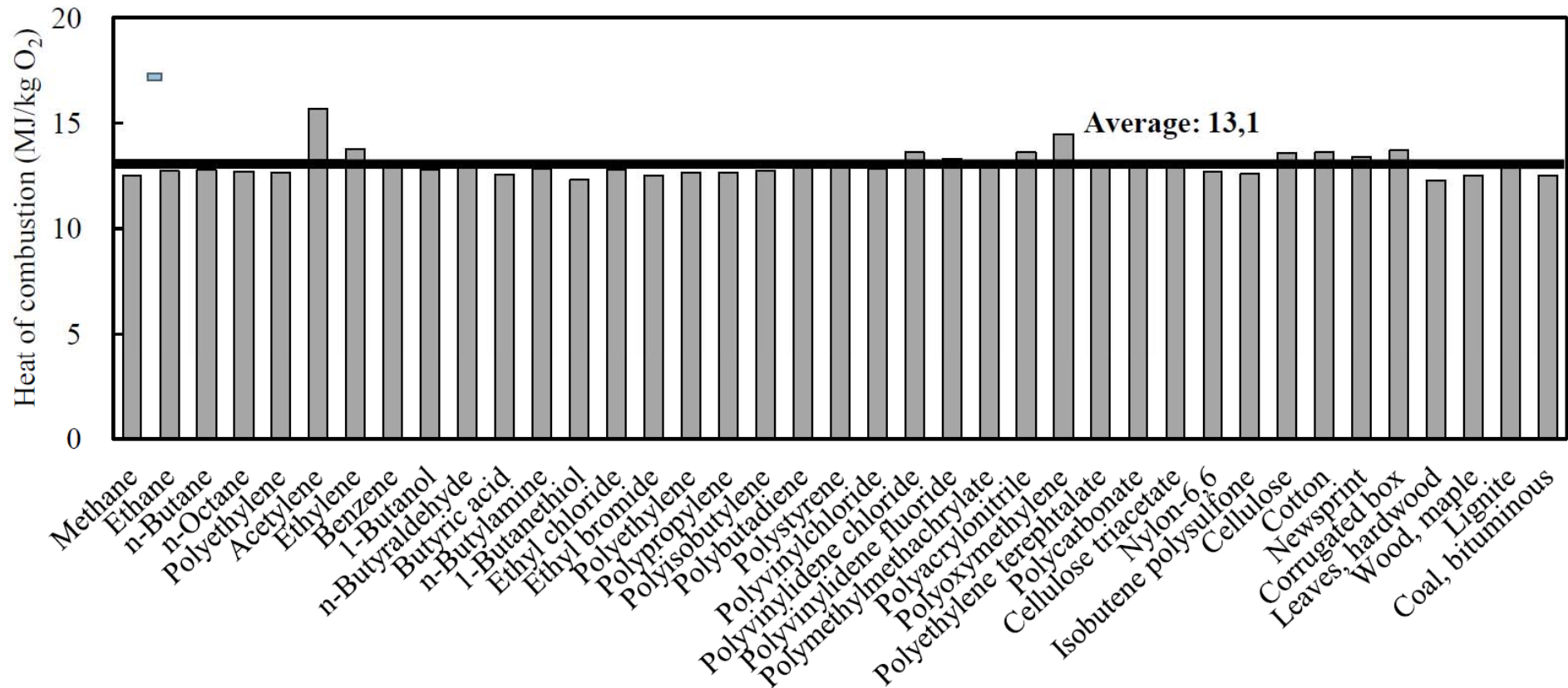
# O<sub>2</sub> Consumption

## COMPLETE COMBUSTION

- Main simplifying assumptions:
  - Energy release per unit mass of O<sub>2</sub>, constant E = 13.1 MJ/kg of O<sub>2</sub> consumed
  - Ideal gas law applies
  - O<sub>2</sub> depletion factor assumes each mole of air required for complete combustion is replaced by 1.105 moles of products

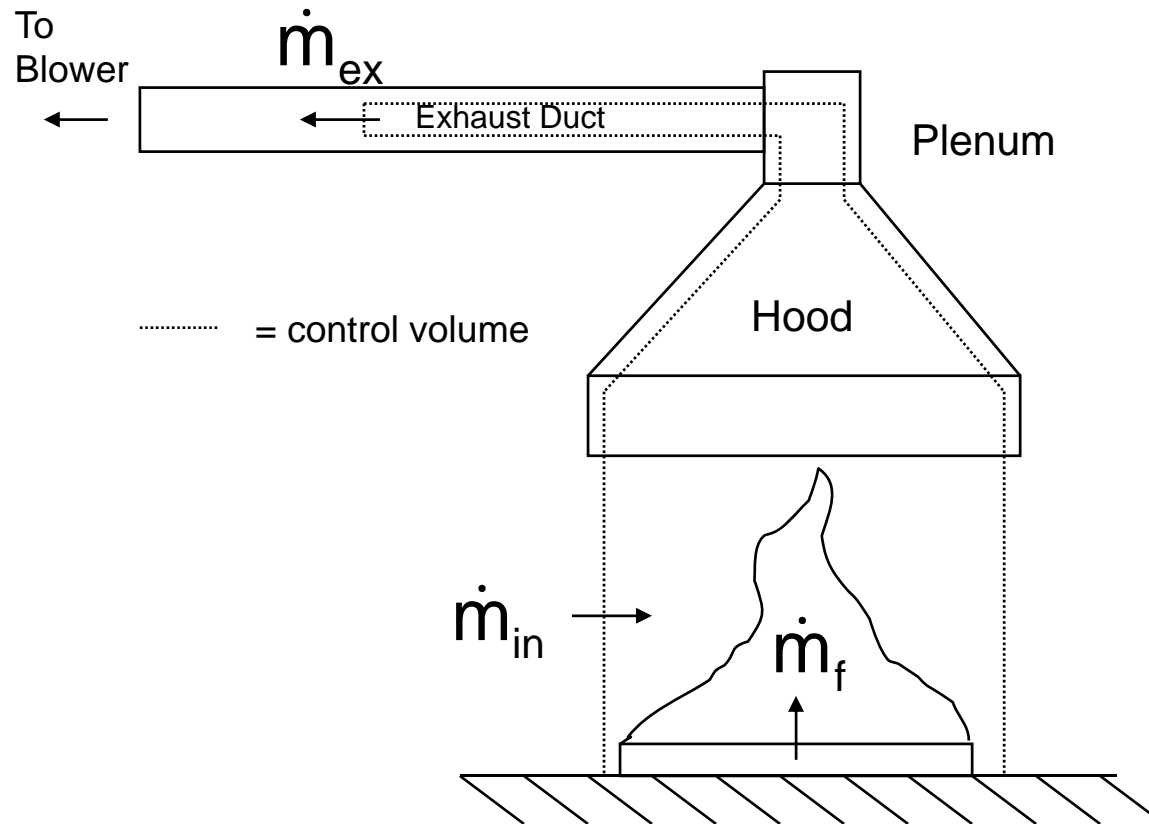
$$n_f + n_{\text{air}} = n_p \approx 1.105n_{\text{air}}$$

# Energy Released per kg of O<sub>2</sub>



HUGGETT, Clayton. 1980. Estimation of rate of heat release by means of oxygen consumption measurements. *Fire and Materials*. 4(2), pp.61-65.

# O<sub>2</sub> Consumption



$$\Delta \dot{m}_{O_2} = \dot{m}_{O_2.in} - \dot{m}_{O_2.ex}$$

# Calculations

- Oxygen concentration is measured at the exhaust  $\Delta\dot{m}_{O_2.ex}$
- Incoming oxygen concentration (air)  $\Delta\dot{m}_{O_2.in}$
- Oxygen consumed :  $\Delta\dot{m}_{O_2}$

$$\Delta\dot{m}_{O_2} = \dot{m}_{O_2.in} - \dot{m}_{O_2.ex}$$



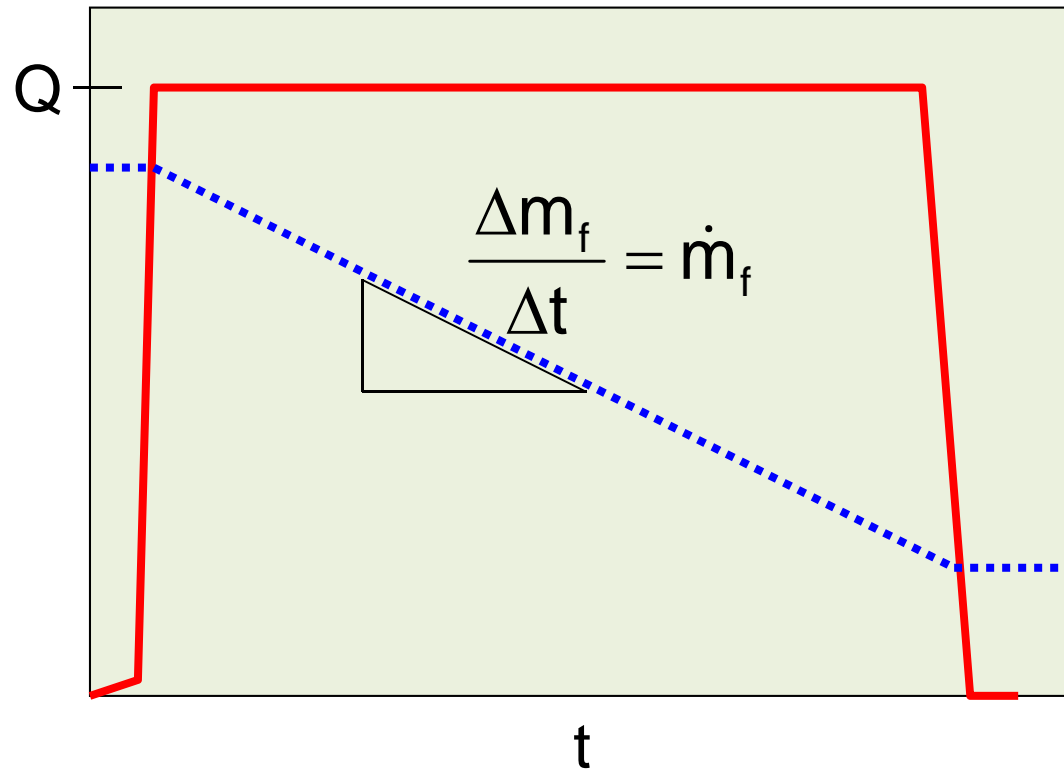
# Calculations

The energy release rate can be calculated as:

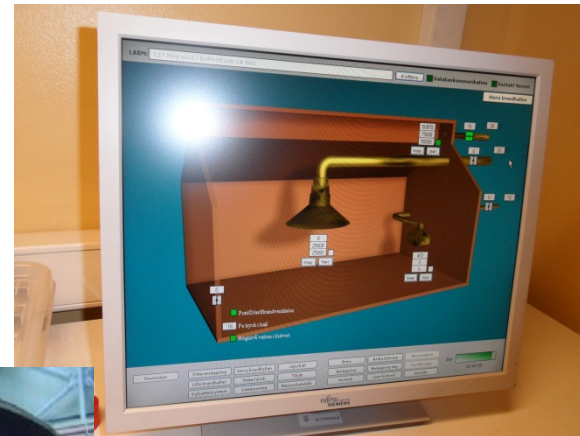
$$Q = 13.1 \Delta \dot{m}_{O_2} \quad [MJ]$$

# Experimental Results

- Ideal Scenario:



# The Real Scale Application- Large Scale Calorimeter



# Aim is

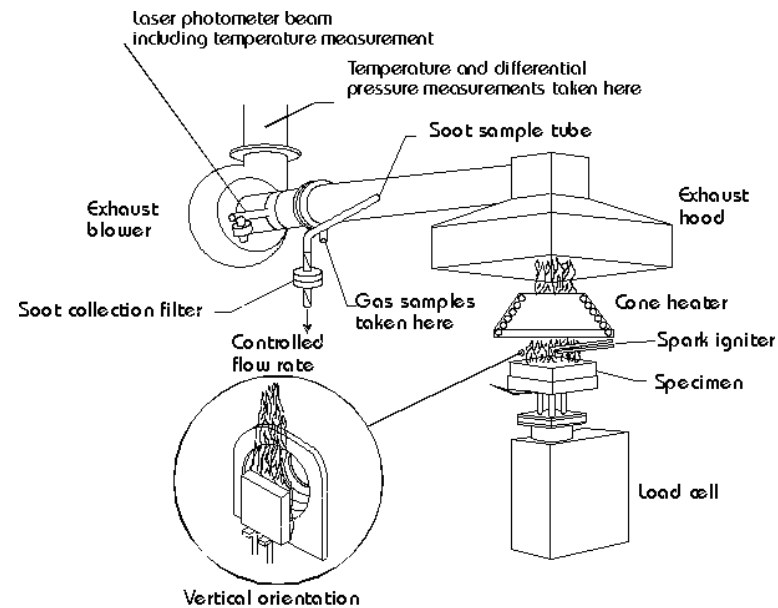
- To introduce the cone calorimeter for determining of
  - the burning rate**
  - Heat Release Rate**
  - time to ignition**

# Aim is

- To get a better understanding of
  - **How Heat Release Rate is measured**
  - **Use of the measured data for calculation**

# Laboratory work – Cone Calorimeter

- Heat Release Rate
- Ignition Temperature
- Partical board (PB) and porous fiberboard (PF)
- Different effect of the incident radiation



# Lab 1. Important information

- *Date – 2<sup>nd</sup> and 3<sup>rd</sup> of October*
- Place – *Complab*. Fire corner.
- Groups of 3-4 people
- Lab report should be written in **ENGLISH** and handed in on the course Fronter page.
- Deadline - **Wednesday the 10th of October**
- Approved report will award to group of students **1.5 point**.

# Groups

https://fronter.com/ltu/main.phtml | Luleå tekniska universitet - Fronter

>> fronter Rooms Tools Online (!) Alexandra Byström

Today S0003B Branddynamik...

You are here: S0003B Branddynamik I > Kursinfo

Rum Deltagare

Kursinfo Kursmaterial Inlämning

Frågor, hjälp Övrigt

Portfolio dolt Lärararkiv dolt

Resource information Expand all

Kursinfo 2010- (Hidden) Recycle Bin

Fronter document

Title: Inlämnings status

Version: Last version: 0.1

Owner: Alexandra Byström

Description:

Last Revised: 2013-09-08 23:27

Insert paragraph Insert image Insert equation

### Inlämnings status S0006B\_Branddynamik I

Namn	Grupp	Inlämning 1	Inlämning 2	Inlämning 3	Labbrapport	Totalt
Anderson, Anne Lee	4					
Andersson, My	4					
Babayan Hezeh Jan, Al	7					
Bergqvist, Emma	2					
Degler, Jakob	6					
Eliasson, Andreas	6					



# Schedule

## – 2<sup>nd</sup> of October.

- Group 1 - 13.00
- Group 3 – 14.00
- Group 4– 15.00

## – 3<sup>rd</sup> of October.

- Group 5 – 10.00
- Group 6– 11.00
- Group 2– 13.00
- Group 7– 14.00
- Group 8– 15.00

# Questions

- What temperature should be set on the cone to get the incident radiation effect at
  - 20 kW/m<sup>2</sup>
  - 40 kW/m<sup>2</sup>?

(According to the calibration data).

# Recordings

Time to ignition	$t_{\text{ign PB}}$	sec
	$t_{\text{ign PF}}$	sec
Samples dimensions (PB and FB)	$L*B*H$	m
Weight of samples	m	kg
Time of extinguished (PB and FB)	$t_{\text{ext, PF}}$	sec
	$t_{\text{ext, FB}}$	
Heat release rate as a function of time	$\dot{Q}$	W/m <sup>2</sup>
Peak rates of heat release (PHRR)	$\dot{Q}_{\text{max}}$	W/m <sup>2</sup>
Time of maximum HRR (PHRR)	$t_{\text{max HRR}}$	sec
Average effective heat of combustion	$\Delta H_{\text{eff, average}}$	MJ/kg
Mass burning rate	$\dot{m}$	g/sec

# Ignition theory

- **piloted ignition** - the surface temperature of 250 °C to 450 °C.
- **auto ignition** - temperature exceeds normally 500 °C.



# Questions

Heat release rate??

Hint: Tsantaridis, L. Reaction to the fire performance of wood and other building products. Doctoral thesis. KTH. 2003

# Report

Short and clear

# Use in your report

- Diagrams
  - Temperature
  - Heat release rate
  - Burning rate on your choice
- Picture from experiment (take a camera)
- Tables
- Ets